

Metal-Free Three-Component Oxyazidation of Alkenes with Trimethylsilyl Azide and N-Hydroxyphthalimide

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Supporting Information

$$R^{1}$$
 R^{2} R^{3} R^{3} R^{2} R^{3} R^{3

ABSTRACT: A novel and facile oxyazidation of alkenes under metal-free and mild conditions has been reported. A remarkable feature of the developed procedure is consecutive construction of C-O and C-N bonds in one step. The process allows quick and selective assembly of alkyl azide from readily available starting materials, where N-hydroxyphthalimide was used as an oxygen-radical precursor and TMSN₃ as the N₃ source. A range of aromatic alkenes bearing synthetically useful functional groups was tolerated.

INTRODUCTION

Nitrogen-containing compounds occupy a significant position in natural products, materials science, and pharmaceuticals, which has inspired the search for milder and more efficient C-N bond-forming reactions. As highly important nitrogencontaining compounds, organic azides have gained great attention and have been recognized as efficient synthetic intermediates, which can be easily transformed into amines under reductive conditions.² The traditional methods for most organic azide syntheses are based on the transformation of alkyl halides via an S_N2 reaction. Recently, chemists found that using azidyl radicals in the functionalization of alkenes can directly realize various straightforward transformations for the construction of substituted organic azides.³ For example, in the past decade, several groups have reported some methods of azidyl radical addition to alkenes followed by formation of C-X (X =O, Se, N, I, F, CN) bonds using TMSN₃, NaN₃, or IN₃ as the N₃ source (Scheme 1, eq 1).⁴ However, to the best of our knowledge, these procedures are not well developed, and up to now, straightforward and efficient approaches for organic azide syntheses by the addition of the oxygen radical to alkenes followed by C-N bond formation have been rarely reported. In addition, direct azidation of alkenes under metal-free oxidation conditions was also developed. 4a,5 For instance, in 2013, the Antonchick^{5a} group independently reported a novel azidation method for the construction of azide oxindoles from alkene compounds using PhI(OCOCF₃)₂ as oxidant (Scheme 1, eq 2). Herein, we report a metal-free functionalization of alkenes by the addition of oxygen radicals followed by trapping with an azidyl source at room temperature to give synthetically useful 2-

Scheme 1. Azidation of Alkenes

Previous work:

$$R^{1}$$
 X N_{3} $X = OR, SeR, N_{3}, I, F, CN (equation 1)$

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Our work:

$$R^{1} \stackrel{\stackrel{\textstyle R^{2}}{\text{\footnote{in}}}}{\mathbb{R}^{1}} R^{3} + \stackrel{\textstyle \mathsf{O}}{\mathbb{R}^{1}} \stackrel{\textstyle \mathsf{N}-\mathsf{OH}}{\mathbb{R}^{1}} \stackrel{\textstyle \mathsf{Phl}(\mathsf{OAc})_{2}}{\mathbb{D}\mathsf{CE},\,\mathsf{RT}} R^{1} \stackrel{\textstyle \mathsf{N}_{3}}{\mathbb{R}^{2}} \stackrel{\mathsf{R}^{2}}{\mathbb{R}^{3}} \stackrel{\mathsf{O}}{\mathbb{R}^{3}} \stackrel{\mathsf{N}_{3}}{\mathbb{R}^{2}} \stackrel{\mathsf{O}}{\mathbb{R}^{3}} \stackrel{\mathsf{N}_{3}}{\mathbb{R}^{2}} \stackrel{\mathsf{O}}{\mathbb{R}^{3}} \stackrel{\mathsf{N}_{3}}{\mathbb{R}^{2}} \stackrel{\mathsf{O}}{\mathbb{R}^{3}} \stackrel{\mathsf{N}_{3}}{\mathbb{R}^{3}} \stackrel{\mathsf{$$

azido-2-(phenylethoxy)isoindolinone compounds, which could provide a broad range of possibilities for further derivatization (Scheme 1, eq 3).

RESULTS AND DISCUSSION

It was revealed that N-hydroxyphthalimide (NHPI) can generate the phthalimide N-oxyl (PINO) radical, which is an active catalytic species to realize C-H bond functionalization.⁶ Recently, PINO was also utilized as a stoichiometric reactant in

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Table 1. Selected List of Screening for the Optimized Conditions^a

entry	1a:2:3	catalyst (%)	oxidant	solvent	yield (%) ^b
1	1:2:1.2	20% CuCl	1.2 equiv of PhI(OAc) ₂	DCE	57
2	2:2:1	10% CuCl₂·2H₂O	1.2 equiv of PhI(OAc) ₂	DCE + 4 Å MS	73
3	2:2:1	5% CuCl ₂ ·2H ₂ O	1.2 equiv of $PhI(OAc)_2$	DCE + 4 Å MS	52
4	2:2:1	_	1.2 equiv of PhI(OAc) ₂	DCE + 4 Å MS	47
5	2:3:1	_	2.0 equiv of PhI(OAc) ₂	DCE + 4 Å MS	70
6	2:3:1	_	2.0 equiv of PhI(OCOCF ₃) ₂	DCE + 4 Å MS	64
7^c	2:3:1	_	2.0 equiv of $PhI(OAc)_2$	DCE + 4 Å MS	20
8	2:3:1	_	2.0 equiv of $PhI(OAc)_2$	$CH_3CN + 4 \text{ Å MS}$	40
9	2:3:1	_	2.0 equiv of PhI(OAc) ₂	DCM + 4 Å MS	30%
10	2:3:1	_	2.0 equiv of PhI(OAc) ₂	THF + 4 Å MS	20
11	2:3:1	_	2.0 equiv of PhI(OAc) ₂	toluene + 4 Å MS	45
12	2:3:1	_	2.0 equiv of PhI(OAc) ₂	acetone + 4 Å MS	56
13	2:3:1	_	_	DCE + 4 Å MS	<5

"Conditions: 3 (0.3 mmol) and 1a (equiv), azide (equiv) in solvent (2.0 mL), at room temperature, in air, 12 h. "Yields of isolated product after chromatography on silica gel. "2 equiv of NaN3 was used instead of TMSN3.

organic synthesis to construct the C-O bond. We envisaged that PINO can be also utilized as a stoichiometric reactant for radical addition, and the results are reported herein. Our investigation began with the reaction of styrene 1a, TMSN₃, and N-hydroxyphthalimide using 20 mmol % CuCl as catalyst and PhI(OAc), as oxidant, and the target product 4a was obtained in 57% yield at ambient temperature (see Table 1, entry 1). To increase the yield, several copper catalysts were evaluated, including CuCl, CuBr, CuI, Cu(OTf)2, CuBr2, Cu(OAc)₂, and CuCl₂·2H₂O, wherein CuCl₂·2H₂O displayed high catalytic activity (see Table S1, entry 18, Supporting Information). When 4 Å MS were added into the reaction system, the yield was increased. To our delight, in the absence of copper catalyst, 4a was still formed (Table 1, entry 4), which meant that the copper catalyst did not play a decisive role in the reaction. When 2.0 equiv of PhI(OAc), was used, the yield increased to 70%, similar to previous results (entry 2 vs entry 5). PhI(OCOCF₃)₂ (2.0 equiv) gave a lower yield (entry 6). When NaN3 was used as the nitrogen source instead of TMSN₃, a very low yield was obtained (entry 7). Meanwhile, screening of solvents revealed that DCE was the best for the reaction under metal-free conditions (entries 8-12). A control experiment showed that only a small amount of product was detected by GC in the reaction in the absence of oxidant $(PhI(OAc)_2)$.

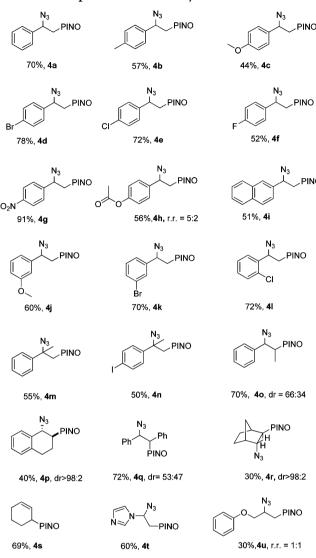
Using the optimized conditions (Table 1, entry 21), the scope of the transformation was investigated. As shown in Scheme 2, a great variety of substituted styrenes were converted to the corresponding alkyl azides in good to moderate yields. Several useful functional groups were tolerated, including fluoro, chloro, bromo, iodo, nitro, ester, ether, and alkyl substituents, at different positions of the styrenes. An electron-withdrawing substituent favored product formation, whereas an electron-donating group gave slightly lower yields. When 4-vinylphenyl acetate was employed, poor regioselectivity was observed (4h, regioisomeric ratio = 5:2). Naphthalenes bearing vinyl groups at position 2 afforded the desired products in 51% (4i) yield. α -Methylstyrene gave the corresponding product 4m in moderate yield; (E)- β -methylstyrene was also compatible

and provided a mixture of diastereoisomers 4o (dr = 66:34). Dihydronaphthalene can also participate in the reaction but gave a low yield (4p). (E)-1,2-Diphenylethene gave a mixture of diastereoisomers 4q (dr = 53:47). Several non-styrenes also reacted. For example, norbornylene can deliver the target product, and only the major isomer was separated despite a low yield (4r). When cyclohexene was employed as a substrate, cyclohexenyl-PINO (4s) was selectively obtained in good yield. A possible mechanism for the formation of cyclohexenyl-PINO product (4s) was proposed (Scheme 3). The imidazole analogue of styrene gave the expected 4t in moderate yield, while allyl phenyl ether gave a mixture of products with no apparent regioselectivity (rr = 1:1, 4u).

As mentioned above, azido compounds are very useful synthetic intermediates and building blocks in organic synthesis. Hence, the azido products were further utilized in other transformations. In the presence of PPh₃ and water, 2-(2-azido-2-phenylethoxy)isoindoline-1,3-dione (4a) was efficiently reduced to the corresponding amine product 5a in 40% yield at room temperature. Product 4a can also undergo the classical click reaction in the presence of CuSO₄·5H₂O (10 mmol %) and Cu powder (4.0 equiv) at room temperature to afford the corresponding triazole 6a in 40% yield (Scheme 4). In the corresponding triazole 6a in 40% yield (Scheme 4).

To gain insight into the mechanism, some control experiments were carried out (Scheme 5). In the absence of TMSN₃ the dioxygenation product 7a was obtained in 68% yield, which was consistent with a mechanism involving the PINO radical intermediate. The structure of the product 7a was further confirmed by X-ray crystallography (see Supporting Information).8 Product 7a can be transformed into 1-phenylethane-1,2diol 8a in moderate yield by cleavage of the N-O bond with Mo(CO)₆ (Scheme 5a). To gain further understanding about the reaction mechanism, inhibition experiments were conducted. When 2.0 equiv of TEMPO (2,2,6,6-tetramethyl-1piperidinyloxy) or BHT (2,6-di-tert-butyl-4-methylphenol) was added into the reaction system, the desired transformation was found to be completely inhibited, providing further evidence that a radical addition mechanism was involved in this transformation (Scheme 5b,c).

Scheme 2. Scope of Metal-Free Oxyazidation of Alkenes^a



"Reaction conditions: alkene (0.6 mmol, 2.0 equiv), TMSN₃ (0.9 mmol, 3.0 equiv), 3 (0.3 mmol), PhI(OAc)₂ (0.6 mmol, 2.0 equiv), DCE (3.0 mL), 4 Å MS,12 h, in air, at room temperature. rr = regioisomeric ratio (major: minor). PINO = phthalimide *N*-oxyl.

Scheme 3. Mechanism for the Formation of 4s

According to the above experimental results and related published research studies, 7c,9 a possible mechanism is outlined in Scheme 6. First, NHPI is oxidized to give the oxygencentered radical PINO. PINO then quickly reacts with styrene to give radical **A**, which can be further oxidized to cation **B** in the presence of PhI(OAc)₂. Last, cation **B** was attacked by N_3^- to give the azido product.

In conclusion, we have developed an efficient metal-free oxyazidation of activated and unactivated alkenes for the synthesis of alkyl azides that holds great potential for the subsequent transformation of the azido unit. A radical addition

Scheme 4. Further Synthetic Transformations

process is involved in this transformation with the formation of C–O and C–N bonds. Further investigations on the scope and synthetic application of these reactions are ongoing in our group.

■ EXPERIMENTAL SECTION

General Remarks. Column chromatography was carried out on silica gel. Unless noted, $^1\mathrm{H}$ NMR spectra were recorded at 400 MHz in CDCl₃ and DMSO- d_6 , and $^{13}\mathrm{C}$ NMR spectra were recorded at 100 MHz in CDCl₃ and DMSO- d_6 . IR spectra were recorded on an FT-IR spectrometer, and only major peaks are reported in cm⁻¹. Melting points were determined on a microscopic apparatus and are uncorrected. All new products were further characterized by HRMS (high resolution mass spectrometry) on a micrOTOF-Q instrument equipped with an ESI source. Copies of $^1\mathrm{H}$ NMR and $^{13}\mathrm{C}$ NMR spectra are provided in Supporting Information. Commercially available reagents 1, 2, and 3 and solvents were used without further purification.

Typical Procedure for the Synthesis of Products 4. To a solution of *N*-hydroxyphthalimide (3, 0.3 mmol, 48.9 mg) in DCE (3.0 mL) were added styrene (1, 0.6 mmol), TMSN₃ (2, 0.9 mmol), PhI(OAc)₂ (193.2 mg, 2.0 equiv), and 4 Å MS (40 mg). The reaction mixture was then stirred for 12 h at room temperature in air. After the reaction, the resulting mixture was quenched with water and extracted twice with EtOAc. The combined organic extracts were washed with brine, dried over Na₂SO₄, and concentrated. Purification of the crude product by flash column chromatography afforded product 4 (petroleum ether/ethyl acetate as eluent (6:1)).

Characterization Data of Compounds 4. 2-(2-Azido-2-phenylethoxy)isoindoline-1,3-dione (4a, 70%, 64.7 mg), mp = 66–68 °C, white solid, 1 H NMR (400 MHz, CDCl₃): 7.68–7.76 (m, 4 H), 7.51–7.52 (m, 2 H), 7.36–7.37 (m, 3 H), 5.50–5.53 (m, 1 H), 3.89–3.94 (dd, 1 H, J = 12.0 Hz, 8.0 Hz), 3.59–3.63 (dd, 1 H, J = 12.0 Hz, 4.0 Hz); 13 C NMR (100 MHz, CDCl₃): 163.3, 135.8, 134.9, 134.4, 134.1, 129.0, 128.5, 127.8, 123.4, 87.5, 54.1; IR (cm⁻¹): 3064, 3035, 2935, 2105, 1789, 1736, 1668, 1459, 1370, 1187, 1126, 1077, 1019, 976, 877, 761, 701; HRMS (ESI) m/z: calcd for $C_{16}H_{12}N_4NaO_3$: M + Na = 331.0807; found: 331.0814.

2-(2-Azido-2-*p*-tolylethoxy)isoindoline-1,3-dione (**4b**, 57%, 55.1 mg), mp = 64–65 °C, white solid, ¹H NMR (400 MHz, CDCl₃): 7.73–7.75 (m, 2 H), 7.67–7.69 (m, 2 H), 7.38–7.40 (m, 2 H), 7.16–7.18 (m, 2 H), 5.47–5.50 (dd, 1 H, J = 8.0 Hz, 4.0 Hz), 3.88–3.93 (dd, 1 H, J = 12.0 Hz, 8.0 Hz), 3.56–3.61 (dd, 1 H, J = 14.0 Hz, 6.0 Hz), 2.32 (s, 3 H); ¹³C NMR (100 MHz, CDCl₃): 163.5, 139.7, 134.5, 132.0, 128.8, 127.9, 123.5, 87.4, 54.2, 21.3; IR (cm⁻¹): 2926, 2103, 1735, 1368, 975, 876, 814, 702; HRMS (ESI) m/z: calcd for $C_{17}H_{14}N_4NaO_3$: M + Na = 345.0964; found: 345.0958.

2-(2-Azido-2-(4-methoxyphenyl)ethoxy)isoindoline-1,3-dione (4c, 44%, 44.6 mg), oil, 1 H NMR (400 MHz, CDCl₃): 7.68–7.76 (m, 4 H), 7.42–7.44 (m, 2 H), 6.87–6.89 (m, 2 H), 5.46–5.49 (m, 1 H), 3.89–3.94 (dd, 1 H, J = 12.0 Hz, 8.0 Hz), 3.79 (s, 3 H), 3.57–3.62 (dd, 1 H, J = 16.0 Hz, 4.0 Hz); 13 C NMR (100 MHz, CDCl₃): 163.4, 160.6, 134.4, 129.4, 128.6, 126.8, 123.4, 113.9, 87.0, 55.1, 53.9; IR (cm $^{-1}$): 2936, 2106, 1735, 1611, 1514, 1369, 1252, 1026, 975, 703;

Scheme 5. Mechanistic Experiments

Scheme 6. Proposed Mechanism

HRMS (ESI) m/z: calcd for $C_{17}H_{14}N_4NaO_4$: M + Na = 361.0913; found: 361.0907.

2-(2-Azido-2-(4-bromophenyl)ethoxy)isoindoline-1,3-dione (4d, 78%, 90.3 mg), mp = 91–92 °C, pale yellow solid, ¹H NMR (400 MHz, CDCl₃): 7.69–7.77 (m, 4 H), 7.49–7.51 (m, 2 H), 7.41–7.43 (m, 2 H), 5.45–5.48 (m, 1 H), 3.85–3.90 (dd, 1 H, J = 14.0 Hz, 4.0 Hz), 3.58–3.62 (dd, 1 H, J = 12.0 Hz, 4.0 Hz); ¹³C NMR (100 MHz, CDCl₃): 163.3, 134.5, 132.1, 131.7, 129.5, 129.4, 128.5, 123.5, 86.8, 53.9; IR (cm⁻¹): 2930, 2105, 1735, 1369, 1285, 1073, 1013, 975, 790, 702; HRMS (ESI) m/z: calcd for $C_{16}H_{11}N_4NaBrO_3$: M + Na = 408.9912; found: 408.9907.

 $2\text{-}(2\text{-}Azido\text{-}2\text{-}(4\text{-}chlorophenyl)ethoxy)isoindoline-1,3-dione (4e, 72%, 73.9 mg), mp = 93–94 °C, pale yellow solid, <math display="inline">^1\text{H}$ NMR (400 MHz, CDCl3): 7.76–7.78 (m, 2 H), 7.70–7.72 (m, 2 H), 7.45–7.49 (m, 2 H), 7.34–7.37 (m, 2 H), 5.47–5.50 (m, 1 H), 3.85–3.91 (dd, 1 H, J=16.0 Hz, 8.0 Hz), 3.58–3.62 (dd, 1 H, J=12.0 Hz, 4.0 Hz); ^{13}C NMR (100 MHz, CDCl3): 163.3, 134.5, 129.2, 128.8, 128.6, 123.5, 86.8, 53.9; IR (cm $^{-1}$): 2931, 2105, 1736, 1671, 1368, 828, 702; HRMS (ESI) m/z: calcd for $C_{16}H_{11}N_4\text{NaClO}_3$: M + Na = 365.0417; found: 365.0412.

 $2\text{-}(2\text{-}\text{Azido-}2\text{-}(4\text{-}\text{fluorophenyl})\text{ethoxy})\text{isoindoline-}1,3\text{-}\text{dione} \ (4f, 52\%, 50.9 mg), mp = 69-70 °C, pale yellow solid, <math display="inline">^1\text{H}$ NMR (400 MHz, CDCl_3): 7.76-7.78 (m, 2 H), 7.70-7.72 (m, 2 H), 7.50-7.52 (m, 2 H), 7.04-7.10 (m, 2 H), 5.47-5.50 (dd, 1 H, J=8.0 Hz, 4.0 Hz), 3.87-3.92 (dd, 1 H, J=14.0 Hz, 6.0 Hz), 3.58-3.63 (dd, 1 H, J=14.0 Hz, 6.0 Hz); ^{13}C NMR (100 MHz, CDCl_3): 163.4, 134.5, 129.9, 129.8, 128.6, 123.5, 115.7, 115.5, 86.8, 54.0; IR (cm $^{-1}$): 2934, 2105, 1736, 1512, 836, 702; HRMS (ESI) m/z: calcd for $C_{16}H_{11}N_4NaFO_3$: M + Na = 349.0713; found: 349.0707.

2-(2-Azido-2-(4-nitrophenyl)ethoxy)isoindoline-1,3-dione (4g, 91%, 96.4 mg), mp = 87–88 °C, pale yellow solid, 1 H NMR (400 MHz, CDCl₃): 8.24–8.26 (m, 2 H), 7.73–7.80 (m, 6 H), 5.56–5.59 (dd, 1 H, J = 4.0 Hz, 4.0 Hz), 3.88–3.93 (dd, 1 H, J = 14.0 Hz, 6.0 Hz), 3.67–3.72 (dd, 1 H, J = 14.0 Hz, 6.0 Hz); 13 C NMR (100 MHz, CDCl₃): 163.3, 148.5, 142.2, 134.8, 128.7, 123.7, 86.6, 53.9; IR (cm⁻¹): 2931, 2107, 1736, 1522, 1348, 853, 702; HRMS (ESI) m/z: calcd for $C_{16}H_{11}N_{5}NaO_{5}$: M + Na = 376.0658; found: 376.0652.

4-(1-Azido-2-(1,3-dioxoisoindolin-2-yloxy)ethyl)phenyl acetate (4h, 56%, rr = 5:2, 61.5 mg), ¹H NMR (400 MHz, CDCl₃): 8.29–8.31 (m, 0.4 H), 8.21–8.23 (m, 0.4 H), 7.90–7.95 (m, 0.5 H), 7.82–7.86 (m, 0.5 H), 7.74–7.76 (m, 2 H), 7.68–7.70 (m, 2 H), 7.54–7.56 (m, 2 H),

7.11–7.14 (m, 2 H), 6.12–6.16 (m, 0.4 H), 5.52–5.55 (m, 1.0 H), 4.28–4.34 (m, 0.4 H), 3.86–3.90 (m, 1.0 H), 3.77–3.80 (m, 0.4 H), 3.58–3.62 (m, 1.0 H), 2.29 (s, 1.2 H), 2.28 (s, 3 H); 13 C NMR (100 MHz, CDCl₃): 169.0, 168.9, 163.3, 159.1, 157.2, 151.5, 151.0, 135.8, 134.4, 134.1, 132.5, 127.9, 123.4, 122.1, 121.7, 86.8, 58.8, 54.1, 50.5, 20.9; IR (cm⁻¹): 2932, 2106, 1738, 1665, 1509, 1370, 1199, 976, 911, 702; HRMS (ESI) m/z: calcd for $C_{18}H_{14}N_4NaO_5$: M + Na = 389.0862; found: 389.0856.

2-(2-Azido-2-(naphthalene-2-yl)ethoxy)isoindoline-1,3-dione (4i, 51%, 54.8 mg), oil, $^1\mathrm{H}$ NMR (400 MHz, CDCl₃): 7.93 (s, 1 H), 7.85–7.87 (m, 1 H), 7.78–7.83 (m, 2 H), 7.66–7.69 (m, 3 H), 7.59–7.61 (m, 2 H), 7.43–7.45 (m, 2 H), 5.68–5.71 (m, 1 H), 3.96–4.02 (dd, 1 H, J=16.0 Hz, 8.0 Hz), 3.64–3.68 (dd, 1 H, J=12.0 Hz, 4.0 Hz); $^{13}\mathrm{C}$ NMR (100 MHz, CDCl₃): 163.4, 134.3, 133.7, 132.7, 132.4, 128.5, 128.1, 127.8, 127.7, 126.7, 126.4, 124.5, 123.4, 87.6, 54.1; IR (cm $^{-1}$): 3058, 2930, 2103, 1734, 1368, 1186, 1126, 975, 751, 702; HRMS (ESI) m/z: calcd for $\mathrm{C}_{20}\mathrm{H}_{14}\mathrm{N}_4\mathrm{NaO}_3$: M + Na = 381.0964; found: 381.0958.

2-(2-Azido-2-(3-methoxyphenyl)ethoxy)isoindoline-1,3-dione (4j, 60%, 60.8 mg), oil, 1 H NMR (400 MHz, CDCl₃): 7.68–7.76 (m, 4 H), 7.24–7.28 (m, 1 H), 7.14–7.15 (m, 1 H), 7.04–7.06 (m, 1 H), 6.87–6.90 (m, 1 H), 5.51–5.54 (m, 1 H), 3.86–3.91 (dd, 1 H, J = 12.0 Hz, 8.0 Hz), 3.83 (s, 3 H), 3.57–3.61 (dd, 1 H, J = 12.0 Hz, 4.0 Hz); 13 C NMR (100 MHz, CDCl₃): 163.3, 159.6, 136.5, 134.4, 129.5, 128.7, 123.4, 119.9, 115.6, 112.7, 87.4, 55.2, 54.3; IR (cm $^{-1}$): 2940, 2104, 1735, 1602, 1368, 1268, 876, 787, 702; HRMS (ESI) m/z: calcd for $C_{17}H_{14}N_4NaO_4$: M + Na = 361.0913; found: 361.0907.

2-(2-Azido-2-(3-bromophenyl)ethoxy)isoindoline-1,3-dione (4k, 70%, 81.1 mg), mp = 65–66 °C, pale yellow solid, 1 H NMR (400 MHz, CDCl₃): 7.77–7.79 (m, 2 H), 7.71–7.73 (m, 2 H), 7.67 (s, 1 H), 7.48–7.52 (m, 2 H), 7.25–7.29 (m, 1 H), 5.42–5.45 (dd, 1 H, J = 8.0 Hz, 4.0 Hz), 3.85–3.90 (dd, 1 H, J = 14.0 Hz, 6.0 Hz), 3.59–3.63 (dd, 1 H, J = 16.0 Hz, 4.0 Hz); 13 C NMR (100 MHz, CDCl₃): 163.3, 137.4, 134.5, 130.1, 128.6, 126.3, 123.6, 122.5, 86.8, 54.1; IR (cm⁻¹): 3065, 2928, 2104, 1737, 1470, 1369, 1188, 1125, 976, 877, 787, 701; HRMS (ESI) m/z: calcd for $C_{16}H_{11}N_4NaBrO_3$: M + Na = 408.9912; found: 408.9907.

2-(2-Azido-2-(2-chlorophenyl)ethoxy)isoindoline-1,3-dione (4l, 72%, 73.9 mg), mp = 60-61 °C, pale yellow solid, ¹H NMR (400 MHz, CDCl₃): 7.97–7.99 (m, 1 H), 7.77–7.79 (m, 2 H), 7.70–7.73 (m, 2 H), 7.37–7.41 (m, 1 H), 7.29–7.33 (m, 2 H), 6.04–6.06 (m, 1 H), 3.80–3.85 (dd, 1 H, J = 14.0 Hz, 6.0 Hz), 3.66–3.70 (m, 1 H); ¹³C NMR (100 MHz, CDCl₃): 163.2, 134.5, 132.9, 132.8, 130.3, 127.1, 123.5, 84.1, 53.4; IR (cm⁻¹): 2928, 2105, 1738, 1367, 975, 876, 759, 702; HRMS (ESI) m/z: calcd for $C_{16}H_{11}N_4NaClO_3$: M + Na = 365.0417; found: 365.0412.

2-(2-Azido-2-phenylpropoxy)isoindoline-1,3-dione (4m, 55%, 53.1 mg), oil, ¹H NMR (400 MHz, CDCl₃): 7.82–7.84 (m, 2 H), 7.74–7.77 (m, 2 H), 7.69–7.71 (m, 2 H), 7.37–7.42 (m, 3 H), 3.86–3.94 (m, 2 H), 1.82 (s, 3 H); ¹³C NMR (100 MHz, CDCl₃): 165.1, 139.3, 134.6, 128.6, 128.2, 126.5, 123.5, 89.6, 58.6, 21.7; IR (cm⁻¹): 3061,

2934, 2104, 1736, 1187, 1115, 972, 767, 702; HRMS (ESI) m/z: calcd for $C_{17}H_{14}N_3N_3O_3$: M + Na = 345.0964; found: 345.0958.

2-(2-Azido-2-(4-iodophenyl)propoxy)isoindoline-1,3-dione (4n, 50%, 67.2 mg), oil, $^1\mathrm{H}$ NMR (400 MHz, CDCl₃): 7.82–7.85 (m, 2 H), 7.73–7.77 (m, 4 H), 7.42–7.44 (m, 2 H), 3.84 (dd, J=12.0 Hz, 32.0 Hz, 2 H), 1.75 (s, 3 H); $^{13}\mathrm{C}$ NMR (100 MHz, CDCl₃): 165.0, 139.3, 137.3, 134.7, 128.9, 128.5, 123.6, 94.8, 89.2, 58.3, 21.7; IR (cm $^{-1}$): 2984, 2105, 1791, 1736, 1368, 1310, 1077, 927, 876, 705; HRMS (ESI) m/z: calcd for $\mathrm{C_{17}H_{13}N_4NaIO_3}$: M + Na = 470.9930; found: 470.9925.

2-(1-Azido-1-phenylpropan-2-yloxy)isoindoline-1,3-dione (4o, 70%, 67.6 mg), dr = 66:34, 1 H NMR (400 MHz, CDCl₃): 7.70-7.72 (m, 2 H), 7.66-7.69 (m, 2 H), 7.52-7.54 (m, 1.3 H), 7.46-7.48 (m, 0.8 H), 7.34-7.39 (m, 3 H), 5.28-5.31 (dd, 1 H, J=8.0 Hz, 4.0 Hz), 4.01-4.08 (m, 1 H), 1.43 (d, 2 H, J=4.0 Hz), 1.10 (d, 1 H, J=8.0 Hz); 13 C NMR (100 MHz, CDCl₃): 163.3, 134.3, 128.7, 128.4, 128.1, 123.3, 91.6, 90.6, 59.8, 59.5, 16.3, 15.9; IR (cm $^{-1}$): 2932, 2107, 1736, 1668, 1374, 983, 876, 701; HRMS (ESI) m/z: calcd for $C_{17}H_{14}N_4NaO_3$: M+Na=345.0964; found: 345.0958.

2-(1-Azido-1,2,3,4-tetrahydronaphthalen-2-yloxy)isoindoline-1,3-dione (4p, 40%, 40.1 mg), dr > 98:2 mp = 104–105 °C, white solid,

¹H NMR (400 MHz, CDCl₃): 7.85–7.87 (m, 2 H), 7.75–7.80 (m, 3 H), 7.25–7.32 (m, 2 H), 7.16–7.18 (m, 1 H), 5.07–5.08 (d, J = 3.6 Hz, 1 H), 4.36–4.37 (m, 1 H), 2.91–2.96 (m, 2 H), 2.55–2.59 (m, 1 H), 2.10–2.16 (m, 1 H); 13 C NMR (100 MHz, CDCl₃): 163.9, 137.3, 134.6, 132.1, 129.5, 126.5, 123.6, 84.2, 58.5, 24.6, 22.9; IR (cm⁻¹): 2932, 2099, 1734, 1370, 1258, 974, 876, 775, 752, 702; HRMS (ESI) m/z: calcd for $C_{18}H_{14}N_4NaO_3$: M + Na = 357.0964; found: 357.0958.

2-(2-Azido-1,2-diphenylethoxy)isoindoline-1,3-dione (4q, 72%, 82.9 mg), mp = 111–113 °C, white solid, dr = 53:47, 1 H NMR (400 MHz, CDCl₃): 7.57–7.59 (m, 2 H), 7.53–7.55 (m, 2 H), 7.47–7.50 (m, 1 H), 7.19–7.39 (m, 5 H), 7.04–7.09 (m, 4 H), 5.57–5.60 (m, 1 H), 5.10–5.12 (m, 0.45 H), 4.99–5.01 (m, 0.51 H); 13 C NMR (100 MHz, CDCl₃): 163.0, 134.1, 128.4, 127.7, 123.1, 123.0, 91.1, 89.5, 68.6, 67.6; IR (cm $^{-1}$): 2927, 2105, 1735, 979, 876, 760, 701; HRMS (ESI) m/z: calcd for $C_{22}H_{16}N_4NaO_3$: M + Na = 407.1120; found: 407.1115.

2-(3-Azidobicyclo[2.2.1]heptan-2-yloxy)isoindoline-1,3-dione (4r, 30%, 26.8 mg, major stereoisomer), mp = 75–76 °C, white solid, $^1\mathrm{H}$ NMR (400 MHz, CDCl₃): 7.84–7.86 (m, 2 H), 7.75–7.77 (m, 2 H), 4.39–4.40 (t, 3J = 2.0 Hz, 1 H), 3.73–3.75 (t, 3J = 2.4 Hz, 1 H), 2.70 (m, 1 H), 2.33–2.35 (m, 1 H), 2.09–2.15 (m, 1 H), 1.71–1.76 (m, 1 H), 1.64–1.67 (m, 1 H), 1.38–1.56 (m, 3 H); $^{13}\mathrm{C}$ NMR (100 MHz, CDCl₃): 163.7, 134.6, 128.8, 123.6, 95.0, 68.5, 42.1, 40.1, 34.7, 26.3, 19.7; IR (cm $^{-1}$): 2962, 2096, 1736, 1372, 1019, 786, 703; HRMS (ESI) m/z: calcd for $\mathrm{C_{15}H_{14}N_4NaO_3}$: M + Na = 321.0964; found: 321.0958.

2-(Cyclohex-2-enyloxy)isoindoline-1,3-dione (4s, 69%, 50.3 mg), mp = 62-63 °C, white solid, ¹H NMR (400 MHz, CDCl₃): 7.83–7.85 (m, 2 H), 7.74–7.77 (m, 2 H), 6.07–6.10 (m, 1 H), 5.92–5.95 (m, 1 H), 4.75–4.76 (m, 1 H), 2.12–2.20 (m, 1 H), 1.99–2.05 (m, 3 H), 1.79–1.87 (m, 1 H), 1.61–1.64 (m, 1 H); ¹³C NMR (100 MHz, CDCl₃): 164.2, 134.9, 134.3, 128.9, 123.8, 123.4, 80.8, 27.2, 25.1, 18.1; IR (cm⁻¹): 2938, 2099, 1734, 1650, 1540, 1518, 790, 702.

2-(2-Azido-2-(1*H*-imidazol-1-yl)ethoxy)isoindoline-1,3-dione (4t, 60%, 53.6 mg), mp = 142–143 °C, white solid, ¹H NMR (400 MHz, DMSO- d_6): 8.63 (s, 1 H), 8.46–8.53 (m, 1 H), 8.08–8.16 (m, 3 H), 7.95–7.97 (m, 1 H), 7.62 (s, 1 H), 6.57–6.60 (m, 1 H), 4.66–4.71 (m, 1 H), 4.50–4.55 (m, 1 H); ¹³C NMR (100 MHz, DMSO- d_6): 168.7, 167.4, 166.7, 137.6, 134.0, 131.8, 129.9, 128.4, 123.6, 118.0, 87.3, 50.9, 21.2; IR (cm⁻¹): 3442, 2113, 1675, 1650, 1026, 1003, 824, 764; HRMS (ESI) m/z: calcd for $C_{13}H_{11}N_6O_3$: M + H = 299.0893; found: 299.0887.

2-(2-Azido-3-phenoxypropoxy)isoindoline-1,3-dione (**4u**, 30%, 30.4 mg), rr = 1:1, ¹H NMR (400 MHz, CDCl₃): 7.71–7.84 (m, 4 H), 7.23–7.28 (m, 3 H), 7.02 (m, 1 H), 6.85–6.88 (m, 0.6 H), 6.73–6.77 (m, 0.4 H), 6.20–6.21 (m, 0.9 H), 5.67–5.72 (m, 0.5 H), 5.51–5.53 (m, 0.6 H), 4.96–4.98 (m, 0.3 H), 4.70–4.72 (m, 0.3 H); ¹³C NMR (100 MHz, CDCl₃): 163.6, 156.5, 150.5, 146.9, 134.4, 134.2, 129.6, 129.4, 128.8, 123.4, 122.5, 117.4, 117.0, 116.2, 114.5, 105.3, 104.3,

103.7, 84.6, 74.8, 70.1, 50.5; IR (cm $^{-1}$): 2935, 2106, 1734, 1493, 1224, 968, 878, 757, 700; HRMS (ESI) m/z: calcd for $C_{17}H_{14}N_4O_4Na$: M + Na = 361.0913; found: 361.0907.

Synthetic Transformation of Product 4a to 5a and 6a. A mixture of 2-(2-azido-2-phenylethoxy)isoindoline-1,3-dione 4a (308 mg, 1.0 mmol) and PPh₃ (786 mg, 3.0 mmol) was placed in a Schlenk tube under air, followed by the addition of THF (2.0 mL) and water (2.0 mL). The reaction was conducted at room temperature for 12 h. After that, 5.0 mL of water was added and the reaction was extracted by ethyl acetate twice (10 mL each time). The crude product was purified by flash column chromatography on silica gel (eluent: petroleum ether/ethyl acetate = 3:1) to afford product 5a (112.8 mg, solid).

2-(2-Amino-2-phenylethoxy)isoindoline-1,3-dione (5a, 40%, 112.8 mg), mp = 113–114 °C, pale yellow solid, $^1\mathrm{H}$ NMR (400 MHz, CDCl_3): 7.79–7.80 (m, 2 H), 7.67–7.68 (m, 2 H), 7.28–7.40 (m, 5 H), 5.19 (s, 2 H, NH_2), 4.91–4.94 (dd, 1 H, J = 4.0 Hz, 4.0 Hz), 4.06–4.09 (dd, 1 H, J = 4.0 Hz, 4.0 Hz), 3.80–3.85 (dd, 1 H, J = 16.0 Hz, 4.0 Hz); $^{13}\mathrm{C}$ NMR (100 MHz, CDCl_3): 167.9, 138.3, 133.7, 131.7, 126.7, 123.0, 83.2, 41.6; IR (cm $^{-1}$): 3473, 3061, 3033, 2940, 1772, 1711, 1424, 1393, 1370, 1188, 1063, 1025, 923, 759, 716; HRMS (ESI) m/z: calcd for $\mathrm{C_{16}H_{15}N_2O_3}$: M + H = 283.1083; found: 283.1097.

A mixture of 2-(2-azido-2-phenylethoxy)isoindoline-1,3-dione 4a (308 mg, 1.0 mmol), ethynylbenzene (122.4 mg, 1.2 mmol), $\rm CuSO_4$: $\rm SH_2O$ (25.0 mg, 0.1 mmol), and Cu powder (256 mg, 4.0 mmol) was placed in a flame-dried Schlenk tube under air, followed by the addition of DMSO (4.0 mL). The reaction was conducted at room temperature for 6 h in dark place. After that, 5.0 mL of water was added and the reaction was extracted by ethyl acetate. The crude product was purified by flash column chromatography on silica gel (eluent: petroleum ether/ethyl acetate = 1:1) to afford product 6a (164 mg, solid).

2-(2-Phenyl-2-(4-phenyl-1H-1,2,3-triazol-1-yl)ethoxy)isoindoline-1,3-dione (6a, 40%, 164 mg), mp = 193–194 °C, pale yellow solid, 1H NMR (400 MHz, CDCl₃): 8.13 (s, 1 H), 7.83–7.84 (m, 2 H), 7.66–7.71 (m, 4 H), 7.47–7.49 (m, 2 H), 7.39–7.43 (m, 3 H), 7.36–7.37 (m, 3 H), 5.79–5.82 (m, 1 H), 5.02–5.07 (m, 1 H), 4.86–4.90 (m, 1 H); 13 C NMR (100 MHz, CDCl₃): 163.3, 147.7, 134.6, 134.1, 130.0, 128.8, 128.7, 128.5, 127.7, 125.7, 123.6, 121.1, 86.8, 53.6; IR (cm⁻¹): 2926, 1734, 1540, 1516, 763, 699; HRMS (ESI) m/z: calcd for $C_{24}H_{18}N_4O_3Na$: M + Na = 433.1277; found: 433.1271.

Typical Procedure for the Synthesis of Product 7a. To a solution of N-hydroxyphthalimide (3, 0.4 mmol, 65.2 mg) in DCE (3.0 mL) were added styrene (1, 0.2 mmol, 20.8 mg), PhI(OAc)₂ (128.8 mg, 0.4 mmol), and 4 Å MS (40 mg). The reaction mixture was then stirred for 12 h at room temperature in air. The resulting mixture was quenched with water and extracted twice with EtOAc. The combined organic extracts were washed with brine, dried over Na₂SO₄, and concentrated. Purification of the crude product by flash column chromatography afforded the product 7a (petroleum ether/ethyl acetate as eluent (3:1), 68%, 58.2 mg).

Compound 7a, mp = 180-181 °C, white solid, ¹H NMR (400 MHz, CDCl₃): 7.66–7.79 (m, 8 H), 7.55–7.58 (m, 2 H), 7.34–7.36 (m, 3 H), 5.85–5.87 (m, 1 H), 4.91–4.95 (m, 1 H), 4.54–4.58 (m, 1 H); ¹³C NMR (100 MHz, CDCl₃): 163.3, 162.9, 134.4, 134.2, 128.6, 128.0, 123.4, 123.3, 85.6, 79.2; IR (cm⁻¹): 2924, 1788, 1733, 1653, 1540, 1462, 1394, 1375, 1187, 1131, 1076, 1021, 972, 878, 700; HRMS (ESI) m/z: calcd for $C_{24}H_{16}N_{2}O_{6}Na$: M + Na = 451.0906; found: 451.0901.

Typical Procedure for the Synthesis of Compound 8a. A mixture of 7a (58.2 mg, 0.136 mmol), $Mo(CO)_6$ (71.8 mg, 0.272 mmol), and Et_3N (206 mg, 2.04 mmol) in $MeCN-H_2O$ (15:1, 3 mL) was stirred at 80 °C for 18 h. Afterward, the mixture was concentrated and the residue purified by chromatography on silica (ethyl acetate/hexane = 1:1) to give 8a (12.2 mg, 65%). ¹ H NMR (400 M H_2) CDCl₃): 7.24–7.28 (m, 5 H), 4.72–4.75 (dd, J = 8.0, 4.0 Hz, 1 H), 4.06 (s, 2 H), 3.55–3.67 (m, 2 H); ¹³C NMR (100 MHz, CDCl₃): 140.5, 128.4, 127.8, 126.0, 74.6, 67.9; the spectral data were identical

to the values of the previous reports: commercially available, CAS no. 16355-00-3.

ASSOCIATED CONTENT

S Supporting Information

Representative experimental procedures, X-ray crystallographic data of 7a, and ¹H NMR and ¹³C NMR spectra of all compounds are available free of charge via the Internet at http://pubs.acs.org.

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Notes

The authors declare no competing financial interest.

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REFERENCES

- (1) (a) Kleemann, A.; Engel, J.; Kutscher, B.; Reichert, D. Pharmaceutical Substances: Synthesis, Patents, Applications, 4th ed.; Thieme: Stuttgart, Germany, 2001. (b) Koch, M. A.; Schuffenhauer, A.; Scheck, M.; Wetzel, S.; Casaulta, M.; Odermatt, A.; Ertl, P.; Waldmann, H. Proc. Natl. Acad. Sci. U. S. A. 2005, 102, 17272. (c) Wang, T.; Jiao, N. Acc. Chem. Res. 2014, 47, 1137.
- (2) (a) McNerny, D. Q.; Mullen, D. G.; Majoros, I. J.; Holl, M. M. B.; Baker, J. R., Jr. Click Chemistry for Biotechnology and Materials Science; Lahann, J.; Ed.; John Wiley & Sons: Chichester, 2009; p 177. (b) Iha, R. K.; Wooley, K. L.; Nyström, A. M.; Burke, D. J.; Kade, M. J.; Hawker, C. J. Chem. Rev. 2009, 109, 5620. (c) Franc, G.; Kakkar, A. K. Chem. Soc. Rev. 2010, 39, 1536.
- (3) (a) Sreekumar, R.; Padmakumarb, R.; Rugmini, P. Chem. Commun. 1997, 1133. (b) Lapointe, G.; Kapat, A.; Weidner, K.; Renaud, P. Pure Appl. Chem. 2012, 84, 1633. (c) Minozzi, M.; Nanni, D.; Spagnolo, P. Chem.—Eur. J. 2009, 15, 7830. (d) Leggans, E. K.; Barker, T. J.; Duncan, K. K.; Boger, D. L. Org. Lett. 2012, 14, 1428. (e) Kapat, A.; Kçnig, A.; Montermini, F.; Renaud, P. J. Am. Chem. Soc. 2011, 133, 13890. (f) Waser, J.; Gaspar, B.; Nambu, H.; Carreira, E. M. J. Am. Chem. Soc. 2006, 128, 11693. (g) Waser, J.; Nambu, H.; Carreira, E. M. J. Am. Chem. Soc. 2005, 127, 8294. (h) Lapointe, G.; Schenk, K.; Renaud, P. Chem.—Eur. J. 2011, 17, 3207. (i) Weidner, K.; Giroult, A.; Panchaud, P.; Renaud, P. J. Am. Chem. Soc. 2010, 132, 17511. (j) Zhang, B.; Studer, A. Org. Lett. 2014, 16, 1790. (k) Zhu, L.; Yu, H.; Xu, Z.; Jiang, X.; Lin, L.; Wang, R. Org. Lett. 2014, 16, 1562. (1) Sequeira, F. C.; Turnpenny, B. W.; Chemler, S. R. Angew. Chem., Int. Ed. 2010, 49, 6365. (m) Li, Z.; Zhang, C.; Zhu, L.; Liu, C.; Li, C. Org. Chem. Front. 2014, 1, 100. (n) Yin, H.; Wang, T.; Jiao, N. Org. Lett. 2014, 16, 2302. (o) Yuan, Y.; Shen, T.; Wang, K.; Jiao, N. Chem.—Asian J. 2013, 8, 2932. (p) Sequeira, F. C.; Chemler, S. R. Org. Lett. 2012, 14, 4482. (q) Dagousset, G.; Carboni, A.; Magnier, E.; Masson, G. Org. Lett. 2014, 16, 4340.
- (4) (a) Zhang, B.; Studer, A. Org. Lett. 2013, 15, 4548. (b) Chung, R.; Yu, E.; Incarvito, C. D.; Austin, D. J. Org. Lett. 2004, 6, 3881. (c) Chouthaiwale, P. V.; Karabal, P. U.; Suryavanshi, G.; Sudalai, A. Synthesis 2010, 3879. (d) Tiecco, M.; Testaferri, L.; Santi, C.; Tomassini, C.; Marini, F.; Bagnoli, L.; Temperini, A. Angew. Chem., Int. Ed. 2003, 42, 3131. (e) Xu, L.; Mou, X.-Q.; Chen, Z.-M.; Wang, S.-H. Chem. Commun. 2014, 50, 10676.
- (5) For selected references, see: (a) Matcha, K.; Narayan, R.; Antonchick, A. P. Angew. Chem., Int. Ed. 2013, 52, 7985. (b) Adamo, I.;

- Benedetti, F.; Berti, F.; Campaner, P. Org. Lett. 2006, 8, 51. (c) Cren, S.; Schär, P.; Renaud, P.; Schenk, K. J. Org. Chem. 2009, 74, 2942. (d) Kapat, A.; König, A.; Montermini, F.; Renaud, P. J. Am. Chem. Soc. 2011, 133, 13890. (e) Angelini, T.; Bonollo, S.; Lanari, D.; Pizzo, F.; Vaccaro, L. Org. Lett. 2012, 14, 4610. (f) Panchaud, P.; Renaud, P. J. Org. Chem. 2004, 69, 3205.
- (6) (a) Ishii, Y.; Nakayama, K.; Takeno, M.; Sakaguchi, S.; Iwahama, T.; Nishiyama, Y. J. Org. Chem. 1995, 60, 3934. (b) Yoshino, Y.; Hayashi, Y.; Iwahama, T.; Sakaguchi, S.; Ishii, Y. J. Org. Chem. 1997, 62, 6810. (c) Hirai, N.; Sawatari, N.; Nakamura, N.; Sakaguchi, S.; Ishii, Y. J. Org. Chem. 2003, 68, 6587. (d) Saha, B.; Koshino, N.; Espenson, J. H. J. Phys. Chem. A 2004, 108, 425. (e) Annunziatini, C.; Gerini, M. F.; Lanzalunga, O.; Lucarini, M. J. Org. Chem. 2004, 69, 3431. (f) Lin, R.; Chen, F.; Jiao, N. Org. Lett. 2012, 14, 4158. (g) Amaoka, Y.; Kamijo, S.; Hoshikawa, T.; Inoue, M. J. Org. Chem. 2012, 77, 9959. (h) Yan, Y.; Feng, P.; Zheng, Q.-Z.; Liang, Y.-F.; Lu, J.-F.; Cui, Y.; Jiao, N. Angew. Chem., Int. Ed. 2013, 52, 5827.
- (7) (a) Hara, T.; Iwahama, T.; Sakaguchi, S.; Ishii, Y. J. Org. Chem. 2001, 66, 6425. (b) Nechab, M.; Einhorn, C.; Einhorn, J. Chem. Commun. 2004, 1500. (c) Lee, J. M.; Park, E. J.; Cho, S. H.; Chang, S. J. Am. Chem. Soc. 2008, 130, 7824. (d) Tan, B.; Toda, N.; Barbas, C. F., III. Angew. Chem., Int. Ed. 2012, 51, 12538. (e) Yao, H.; Yamamoto, K. Chem.—Asian J. 2012, 7, 1542. (f) Ghosh, R.; Olofsson, B. Org. Lett. 2014, 16, 1830. (g) Petrassi, H. M.; Sharpless, K. B.; Kelly, J. W. Org. Lett. 2001, 3, 139.
- (8) CCDC 1026407 (7a) contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Center via www.ccdc.cam. ac.uk/data request/cif.
- (9) (a) Antonchick, A. P.; Burgmann, L. Angew. Chem., Int. Ed. 2013, 52, 3267. (b) Matcha, K.; Antonchick, A. P. Angew. Chem., Int. Ed. 2013, 52, 2082.